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Modern Chemistry Chapter 8. A number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. When the positive ion of one compound replaces the positive ion of another to form two new compounds.

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A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left

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to its representation on the right. c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX + A$ a double-displacement (d) $AX + A \rightarrow X$

8 Chemical Equations and Reactions

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equation (8.1) a way to describe a chemical reaction using
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