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B) CH_2Cl_2 C) $\text{ClHC}=\text{CH}_2$ D) *trans*- $\text{ClHC}=\text{CHCl}$ E) None of
these choices. Answer: D Topic: Molecular geometry, polarity
Section Reference 1: 1.5 and 2.3 Difficulty: Medium 12)

Which molecule would have a dipole moment greater than
zero? A) BeCl_2 B) BCl_3 C) CO_2 D) H_2O E) CCl_4 Answer: D

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