

## Sweet 16 Chemistry Compound Tournament Answers

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This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important topics.

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Sweet 16 Chemistry Ion Tournament

How to do Sweet 16 chemistry Ion Tournament? The rule is to fill out the tournament brackets. For the first round you just add charge then. The second round combine the two ions.

How to do Sweet 16 chemistry Ion Tournament? | Yahoo Answers

Sweet 16 Chemistry of Gases Tournament? First round: Predict the Gaseous product. Second round: Predict the one that dissolves in water to give an acidic solution. Semis: The gas that is more dense. Finals: The winner is a greenhouse gas produced in volcanic emissions. First round: Sodium Carbonate vs. Hydrochloric acid...

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Sweet 16 Chemistry Compound Tournament Answer

Many tournaments and field trips have allowed Kate and I to both do some long term substitution. On one of these weeks, I had a lower end third form class (9th grade). We were reading Trifles, a play about a woman who kills her husband in rural Iowa, by Susan Glaspell. I love this play, and wanted to get the students involved too.

Katie In Oz — LiveJournal

The winner of the tournament is the compound in the finals that has the larger molar mass. Two versions of the Sweet 16 Chemistry Compound Tournament are enclosed. One uses traditional chemical names [e.g. ferrous, cupric, etc] and the other uses the Stock system for chemical names [iron(II), copper(II)].

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